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Asymmetric Synthesis of Phosphinous Tungsten Complexes

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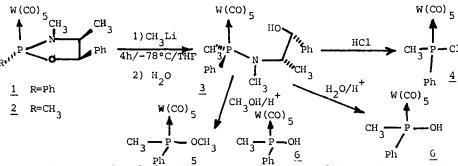
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Asymmetric Synthesis of Phosphinous Tungsten Complexes

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The asymmetric synthesis of phosphinous compounds from diheterophosphacycloal kane-1,3,2 was investigated in a transition metal complex series. Complexes 1 and $\underline{2}$ were prepared from diaminophosphine, (-)-ephedrine and W (CO)₅ THF ($\underline{1}$: δ^{31} P=+147,6 ppm; $J_{pw}=313 \text{ Hz M}^{+}=595$; $2 \delta^{3}P=+156 \text{ ppm}$ $J_{pw}=304 \text{ Hz M}^{+}=533$; 90%dp) in two steps. Methyl lithium reacted with $\underline{1}$ to give stereospecific $\underline{3}$ (80%Yield) by P-O linkage $(\frac{3}{2})^{31}$ P=+64 ppm; J_{pw} =261 Hz; M⁺-28=583). The aminophosphine complex 3 was stereochemically stable and was used for studies of synthetic applications. HCL gas reacted with $\frac{3}{2}$ in CH_2Cl_2 to give the non optically active chlorophosphine complex $\underline{4}$ (δ^{31} P=+103,6 ppm; J_{DW} =290 Hz; M^{*+} =482). This compound immediately gave salt $\underline{6}$ (δ^{31} p=+66,7 ppm; J_{pm} =240 Hz) by reaction with (-)-menthol and triethylamine. The acid methanolysis of 3 gave a mixture of $\underline{5}$ and $\underline{6}$ and the unchanged (-)-ephedrine salt $\underline{5}$:30% yield; δ^{31} P=+114 ppm; $J_{PW} = 280 \text{ Hz}$; $\left[\alpha\right]_{D} = +1.2^{\circ} (CH_{2}Cl_{2})$; $M^{*+} = 478$; $\frac{6}{6} : 60\% \text{ yield}$; $\delta^{31}P = +102.9 \text{ ppm}$; $J_{pw}=264 \text{ Hz}$; [α] =+16,9° (CH₂Cl₂); M⁺=464]. Compound 6 was thus obtained with a 80% yield and a specific rotation of + 20,2° (CH₂Cl₂) in isopropanol/H₂SO₄ 5M medium. The enantiomeric excess of 6 was determined by RMN³¹P. Acid hydrolysis of 3 or the reaction with CH_SO_H, gave phosphinous acid complex 6 with an optical rotation of + 4,8° or-1,8° respectively. These results provide precious informations about the stereochemistry and reactivity of the P-N linkage in this aminophosphine transition metal series, which differs notably from that of the corresponding (PO) N bond.